

## Cobalt Analysis

### Procedure for Preparation of CoEDTA for Use as an Indigestible Marker

#### Materials:

Large beakers, 4 liters  
Graduated cylinders, 1,000 ml  
Stirring-hotplate  
Large Buchner funnel  
Whatman filter paper (fast) or white coffee filters

#### Reagents:

**Cobalt II acetate.4H<sub>2</sub>O** (Mallinckrodt Cat # 04519, FW 249.10, CAS # 6147-53-1 or Baker Cat # 1658-01, FW 249.10, CAS # 6147-53-1)

**Ethylenediaminetetraacetic acid** (acid form) (Sigma-Aldrich Cat # ED-1KG, FW 292.24, CAS # 60-00-4)

**Lithium hydroxide (LiOH.H<sub>2</sub>O)** (Fisher Cat # L127-500, FW 41.96, CAS # 1310-66-3)

**30% Hydrogen peroxide solution** (VWR Cat # JT2190-3, FW 34.02, CAS # 7722-84-1)

**95% Ethanol** (CAS # 64-17-5)

**80% Ethanol**, dilute 800 ml 95% ethanol (CAS # 64-17-5) to 1 liter with water

#### Procedure:

The following amounts of reagents represent a triple batch of CoEDTA and should be weighed into a 4 liter beaker.

1. Weigh 75 g cobalt II acetate, 87.6 g EDTA, and 12.9 g lithium hydroxide.H<sub>2</sub>O (7.35 g anhydrous form) into a 4 liter beaker.
2. Add 600 ml dd water and dissolve with heating if necessary.
3. Cool and add 60 ml of 30% hydrogen peroxide solution.
4. Allow to stand 2-3 hours or overnight.
5. Add 900 ml of 95% ethanol and refrigerate overnight (loosely covered).
6. Filter with vacuum and wash crystals with about 3 liters of 80% ethanol.
7. Dry crystals in 60 °C oven and store dry crystals in a brown glass jar until ready to re-suspend.
8. Re-suspend crystals in dd water (2 g per 100 ml dd water) and measure cobalt concentration on an atomic absorption spectrophotometer.

## **Procedure for Extracting and Measuring Cobalt in Feces, Rumen Fluid or Dose**

### **Materials:**

30-50 ml screw cap tubes  
Whatman #1 filter paper  
Centrifuge  
Pipettes  
Atomic Absorption Spectrophotometer  
Shaker

### **Reagents:**

#### **0.05 M EDTA (ethylenediaminetetraacetic acid) solution:**

Weigh 58.4 g EDTA (Sigma-Aldrich Cat # ED-1KG, FW 292.24, CAS # 60-00-4) and 15.3 g KCl (potassium chloride) (Mallinckrodt Cat #6838-05, FW 74.55, CAS # 7447-40-7 or EM Cat # PX1405-5, FW 74.55, CAS # 7447-40-7); place both in a 4 liter volumetric flask. Add about 3,900 ml dd water, stirring to dissolve reagents, and adjust pH to 6.5 with concentrated reagent grade ammonium hydroxide (Fisher Cat # A470-500, FW 35.05, CAS # 1336-21-6). Add water to a total volume of 4 liters.

**Cobalt Standards** (Perkin Elmer Cat #N9303766)

**18 M $\Omega$  water**

### **Extraction of Cobalt in Feces:**

1. Weigh out 0.2 g dried, ground (2 mm screen) fecal sample in duplicate and place in 30-50 ml screw cap tube.
2. Add 20 ml of the 0.05 M EDTA solution.
3. Cap the tubes and shake for 30 minutes.
4. Filter sample through a Whatman #1 filter paper, twice.
5. Read concentration and/or absorbency of cobalt on the atomic absorption spectrophotometer.

### **Cobalt Measurement of Rumen Fluid or Dose Sample:**

1. Thaw rumen fluid in a plastic container.
2. Centrifuge rumen fluid at 16,000 to 18,000 x g.
3. Dilute supernatant to an appropriate range for measurement by atomic absorption spectrophotometer.

Example:

| <u>Post-dosing Hours</u> | <u>Rumen Fluid, ml</u> | <u>H<sub>2</sub>O, ml</u> | <u>Dilution factor</u> |
|--------------------------|------------------------|---------------------------|------------------------|
| 1-8                      | 1                      | 4                         | 5                      |
| 8-14                     | 2.5                    | 2.5                       | 2                      |
| 15+                      | 5                      | 0                         | 1                      |

4. The AA spec is maximized with a 5 ppm standard, wavelength of 240.7, and slit of 0.2 nm. The working standard curve used is 0.5, 1, 2, 3, 4, 5 ppm and is diluted in 18 MOhm water.

### Reference

E. C. Prigge and G. Varga, West Virginia, University, Uden et al. (1980) J. Sci. Food Agr. 31:625.

Hart, S. P. and C. E. Polan. 1984. Simultaneous extraction and determination of ytterbium and cobalt ethylenediaminetetraacetate complex in feces. J. Dairy Sci. 67:888-892.